Dilute magnetic semiconductors

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 Spintronics (a neologism meaning "spin transport electronics"), also known as magnetoelectronics or spin electronics.







Primary requirements for spintronic materials:

- Room temperature ferromagnetism.
- Compatible with electronic devices.
- Semiconducting and ferromagnetic properties should coexist at room temperature.

Dilute Magnetic Semiconductors (DMS) Nonmagnetic Ferromagnetic Semiconductor DMS Crystal Add TM Ions Mn, Fe, Co

GaAs, ZnO, AlN, TiO₂

Typical representatives. - Mn-doped GaAs - Mn-doped ZnO

When the magnetic coupling occur in the doped

semiconductors?



Ferromagnetism in bulk Mn-doped ZnO

- No trace of any secondary phase or other impurities was found.
- Room temperature ferromagnetism discovered in Mn doped ZnO [1].

Hysteresis curve at Room temperature for 2 at. % Mn-doped ZnO



In general, the solid solubility of transition metals in semiconductors is low.

1. P. Sharma, A. Gupta, K. V. Rao et. al. Nature Materials 2, p. 673 - 677 (2003)

Type of impurity source of magnetism in DMS systems

- Clusters \ Segregations
- Secondary Phases (Spinels)
- Extrinsic impurity



SQUID magnetometer
 (superconducting quantum interference devices)

Clusters



 Nanoclusters of the TM ion form the most obvious candidates for impurity sources of magnetism.

Secondary Phases (Spinels)

Hysteresis curve for ZnCo₂O₄ phase in the Co-doped ZnO



general formula



3. X. Wang, J. Xu, X. Yu, et. al., Applied Physics Letters 91, p. 031908, 2007

Extrinsic impurities

Magnetic moment of small amount extrinsic impurity is

$$\mu = 5 * 10^{-6} emu$$

- Abraham undertook a systematic study of the growth of HfO₂ thin films [2].
- No ferromagnetic signal
 Observation of a magnetic signal
 Teflon tweezers
 Stainless steel tweezers
- 2. D.W. Abraham, Frank, MM, Guha, S, Appl. Phys. Lett. 87, p. 252502, 2005

Metal doped In₂O₃

- 1. Nanocrystalline In₂O₃ is diamagnetic.
- 2. Bulk TM-doped (M = Cr, Mn, Fe, Ni, Cu) In_2O_3 is intrinsically paramagnetic, with a paramagnetic effective moment originating from the dopant.
- 3. Magnetic behaviour in bulk material are linked to the presence of small amounts of magnetic secondary phases in the samples.

Magnetic phenomena in DMS have been attributed to experimental artefact, such as influence of synthesis conditions or measurements errors.

 D. Berardan, E. Guilmeau, D. Pelloqun et. al. Journal of Magnetism and Magnetic Materials 320, p. 983-989, 2008.

$Mn - doped In_2O_3$

Hysteresis loop in Mn-doped In₂O₃



XRD spectrum from the sample showing the presence of oriented Mn_2O_3 fractions.



4. Wakano T., Fujimura N., Morinaga Y., Abe N., Ashida A. Physics E 10, 260, 2001 12

Transmission electron microscope (TEM)



 The dilution of TM ions used to dope these DMS materials means that it is quite conceivable that the cross section does not contain any such impurities.[6]

 O. Kazakova, R. Morgunov, J. Kulkarni, et. Al. Physical Review B (Condensed Matter and Materials Physics) 77, p. 235317, 2008.

X-Ray Absorption Spectroscopy (XAS)



Fe doped BaTiO₃



 BaTiO₃ – ferroelectric material with perovskite structure

Doping with iron:

 $Ba^{2+}(Ti_{1-x}^{4+} Fe_x^{3+})O_{3-x/2}^{-2}$

aliovalent substition

Room temperature ferromagnetism in Fe doped BaTiO₃



8. Fangting Lin, Dongmei Jiang, et al., Phisica B 403, p. 2525-2529, 2008.
 9. Sugata Ray, Priya Mahadevan, et al. Physical Review B 77, p. 104416 (2008)

Room temperature ferromagnetism

in Fe doped BaTiO₃

Magnetic hysteresis loop of the BaTi_{0.95}Fe_{0.05}O₃ ceramics at room temperature



(a) High-resolution transmission electron microscopy of the $BaTi_{0.95}Fe_{0.05}O_3$ ceramics, and selected area electron diffraction in the inset; (b) transmission electron microscopy the $BaTi_{0.95}Fe_{0.05}O_3$ ceramics;



10. B. Xu, K. B. Yin, J. Lin, Y. D. Xia, et al., Physical Review B 79, p. 134109, 2009.¹⁷

Our study: Fe-doped BaTiO3

The purpose of this research is detailed studies on the existence of the intrinsic magnetic ordering in Fe doped BaTiO₃.

Our study: Fe-doped BaTiO3

• Sample preparation:

- Solid state reaction method.
- Ba(Ti_{1-x}Fe_x)O_{3-x/2}, where x=2, 10 % of Fe –synthesis in nitrogen 1250°C/5h.
- $Ba(Ti_{1-x}Fe_x)O_{3-x/2}$, where x=2, 10 % of Fe –synthesis in oxygen 1250°C/5h.
- X-Ray Diffraction (XRD).
- X-Ray Absorption Spectroscopy (XAS).
- Magnetic measurements.

XRD patterns



Fe K-edge XANES spectra in Ba(Ti_{0.98}Fe_{0.02})O_{2.99}



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Valence state of Fe in Fe doped BaTiO₃



Fe K- edge XANES spectra of $Ba(Ti_{0.98}Fe_{0.02})O_{2.99}$ (blue line – annealed in nitrogen, red line – in oxygen) and reference samples (light violet line – FeSO₄, green line – Fe₂O₃) with well known Fe valence state. 22

EXAFS analysis



The EXAFS spectra $\chi(k)$ of Fe in Ba(Ti_{0.98}Fe_{0.02})O_{2.99} (blue line – annealed in nitrogen, red line – in oxygen)



$$\chi(k) = \sum_{i} A_{i}(k) \sin(2kR_{i} + \delta_{i}),$$



• with the atom-specific phase shift δ_i and the amplitude factor

$$A_{i}(k) = \frac{N_{i}}{kR_{i}^{2}} S_{0}^{2} F_{i}(k) e^{(-2k^{2}\sigma_{i}^{2})} e^{-R_{i}\lambda_{i}}$$



Red line - Fe K-edge Fourier transforms of Ba(Tio.98Feo.o2)O2.99 annealed in oxygen.
Blue line - Fe K-edge Fourier transforms of Ba(Tio.98Feo.o2)O2.99 annealed in nitrogen.

EXAFS analysis of Fe doped BaTiO₃



Blue line - Fe K-edge Fourier transforms of $Ba(Ti_{0.98}Fe_{0.02})O_{2.99}$ (experiment). Red line - model of the undistorted $BaTiO_3$ structure around Ti site. Green line - model of the undistorted $BaTiO_3$ structure around Ba site. 26

EXAFS analysis of Fe doped BaTiO₃



Туре	Number	Distance, A	Debye-Waller factor
0	4	2.05(1)	0.006(4)
0	2	1.90(3)	0.006 (4)
Ва	8	3.44(1)	0.011(1)
Ti	4	4.00(2)	0.012(2)

Туре	Number	Distance, A	Debye-Waller factor
0	4	2.05(1)	0.004(2)
0	2	2.011(8)	0.0014
Ва	8	3.46(1)	0.006
Ti	4	3.98(2)	0.009

Magnetic measurements



Conclusion

- We synthesized 4 samples of Fe-doped BaTiO3 by solid state reaction method.
- No magnetic ordering exists in bulk Fe-doped BaTiO₃.
- All iron in Ba(Tio.98Feo.o2)O2.99 samples is in trivalent form and Fe3+ atoms substituted Ti4+ on B sites of the perovskite BaTiO3 host matrix.
- No impurities or secondary phases were observed in well-processed Ba(Tio.98Feo.o2)O2.99.

Thank you for your attention!!!